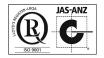


CERTIFICATE OF ANALYSIS FOR

Copper-Cobalt Ore (Democratic Republic of the Congo) CERTIFIED REFERENCE MATERIAL OREAS 556



Document: COA-1475-OREAS 556-R1

TABLE OF CONTENTS

INTRODUCTION	8
SOURCE MATERIAL	8
PERFORMANCE GATES	9
COMMINUTION AND HOMOGENISATION PROCEDURES	9
PHYSICAL PROPERTIES	9
ANALYTICAL PROGRAM	10
STATISTICAL ANALYSIS	10
Homogeneity Evaluation	16
PARTICIPATING LABORATORIES	17
PREPARER AND SUPPLIER	22
METROLOGICAL TRACEABILITY	22
COMMUTABILITY	22
INTENDED USE	23
STABILITY AND STORAGE INSTRUCTIONS	23
INSTRUCTIONS FOR CORRECT USE	24
HANDLING INSTRUCTIONS	25
LEGAL NOTICE	25
DOCUMENT HISTORY	25
QMS ACCREDITATION	25
CERTIFYING OFFICER	25
REFERENCES	26
APPENDIX	26
LIST OF TABLES	
Table 1. Certified Values and Performance Gates for OREAS 556	3
Table 2. Indicative Values for OREAS 556.	
Table 3. Physical properties of OREAS 556.	9
Table 4. 95% Confidence & Tolerance Limits for OREAS 556	11
LIST OF FIGURES	
Figure 1. Co by 4-acid digestion in OREAS 556	18
Figure 2. Co by Peroxide Fusion in OREAS 556	19
Figure 3. Cu by 4-acid digestion in OREAS 556	20
Figure 4. Cu by Peroxide Fusion in OREAS 556	21

Table 1. Certified Values and Performance Gates for OREAS 556.

								I UREAS			
Constituent	Certified		Absolute	Standard	Deviations	3	Relative	Standard D	eviations	5% w	rindow
Consuluent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High
Borate Fusion	1 XRF										
Al ₂ O ₃ , wt.%	9.43	0.127	9.18	9.69	9.05	9.81	1.35%	2.69%	4.04%	8.96	9.90
BaO, ppm	576	51	473	678	421	730	8.93%	17.85%	26.78%	547	604
CaO, wt.%	3.86	0.084	3.69	4.03	3.61	4.11	2.17%	4.34%	6.52%	3.67	4.05
Co, wt.%	2.10	0.048	2.00	2.20	1.96	2.24	2.27%	4.54%	6.81%	1.99	2.20
Cr ₂ O ₃ , ppm	106	13	81	131	68	144	11.88%	23.77%	35.65%	101	111
Cu, wt.%	2.18	0.070	2.04	2.32	1.97	2.39	3.21%	6.42%	9.63%	2.07	2.29
Fe ₂ O ₃ , wt.%	1.18	0.028	1.12	1.24	1.09	1.26	2.42%	4.83%	7.25%	1.12	1.24
K ₂ O, wt.%	2.15	0.042	2.07	2.23	2.02	2.28	1.96%	3.92%	5.87%	2.04	2.26
MgO, wt.%	5.58	0.061	5.46	5.70	5.40	5.76	1.10%	2.20%	3.30%	5.30	5.86
MnO, wt.%	0.089	0.006	0.076	0.101	0.070	0.107	6.91%	13.81%	20.72%	0.084	0.093
P ₂ O ₅ , wt.%	0.113	0.006	0.102	0.124	0.096	0.130	5.03%	10.06%	15.08%	0.107	0.119
SiO ₂ , wt.%	60.82	0.715	59.39	62.24	58.67	62.96	1.17%	2.35%	3.52%	57.77	63.86
SO ₃ , wt.%	4.22	0.072	4.08	4.37	4.01	4.44	1.70%	3.40%	5.10%	4.01	4.44
SrO, ppm	84	22	40	128	18	150	26.04%	52.08%	78.12%	80	88
TiO ₂ , wt.%	0.577	0.015	0.547	0.606	0.533	0.620	2.53%	5.05%	7.58%	0.548	0.605
Thermogravin	netry										•
LOI ¹⁰⁰⁰ , wt.%	7.27	0.607	6.06	8.49	5.45	9.10	8.35%	16.70%	25.05%	6.91	7.64
Infrared Comb	bustion	l			l						
C, wt.%	1.80	0.031	1.74	1.86	1.71	1.89	1.74%	3.48%	5.22%	1.71	1.89
S, wt.%	1.69	0.042	1.60	1.77	1.56	1.81	2.49%	4.97%	7.46%	1.60	1.77
Sulphuric Aci	d 5% Leach								•		•
Co, wt.%	0.823	0.086	0.650	0.995	0.564	1.081	10.49%	20.98%	31.47%	0.781	0.864
Cu, wt.%	0.862	0.072	0.719	1.005	0.647	1.077	8.31%	16.62%	24.93%	0.819	0.905
Sulphuric Aci	d 10% Leac	h							•		•
Co, wt.%	0.828	0.067	0.694	0.963	0.627	1.030	8.11%	16.22%	24.34%	0.787	0.870
Peroxide Fusi	ion ICP										•
Ag, ppm	< 1	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND
Al, wt.%	4.95	0.139	4.67	5.23	4.53	5.37	2.82%	5.63%	8.45%	4.70	5.20
As, ppm	30.0	4.3	21.3	38.7	17.0	43.1	14.47%	28.94%	43.41%	28.5	31.5
B, ppm	475	23	429	521	407	544	4.81%	9.62%	14.43%	451	499
Ba, ppm	501	24	453	549	429	573	4.76%	9.52%	14.29%	476	526
Bi, ppm	4.66	0.437	3.78	5.53	3.35	5.97	9.39%	18.78%	28.17%	4.42	4.89
Ca, wt.%	2.80	0.086	2.63	2.97	2.54	3.05	3.07%	6.14%	9.21%	2.66	2.94
Ce, ppm	83	2.2	79	88	77	90	2.62%	5.25%	7.87%	79	88
Co, wt.%	2.13	0.044	2.04	2.22	2.00	2.26	2.06%	4.12%	6.17%	2.02	2.24
Cr, ppm	78	15	49	108	35	122	18.55%	37.10%	55.65%	75	82
Cs, ppm	2.09	0.180	1.73	2.45	1.55	2.63	8.61%	17.22%	25.83%	1.98	2.19
Cu, wt.%	2.20	0.059	2.08	2.32	2.02	2.38	2.68%	5.37%	8.05%	2.09	2.31
Dy, ppm	4.51	0.340	3.83	5.19	3.49	5.53	7.55%	15.09%	22.64%	4.28	4.73
Er, ppm	2.74	0.225	2.29	3.19	2.06	3.41	8.21%	16.42%	24.63%	2.60	2.87
Eu, ppm	0.80	0.09	0.62	0.97	0.53	1.06	10.98%	21.96%	32.95%	0.76	0.84
Fe, wt.%	0.836	0.033	0.771	0.901	0.738	0.934	3.90%	7.81%	11.71%	0.794	0.878
Ga, ppm	14.0	1.7	10.7	17.4	9.0	19.1	11.97%	23.94%	35.91%	13.3	14.7
Gd, ppm	4.85	0.62	3.60	6.09	2.98	6.72	12.85%	25.69%	38.54%	4.61	5.09
SLunit equival											L

SI unit equivalents: ppm, parts per million \equiv mg/kg \equiv μ g/g \equiv 0.0001 wt.% \equiv 1000 ppb, parts per billion.

Note: intervals may appear asymmetric due to rounding; the number of decimal places quoted does not imply accuracy of the certified value to this level but are given to minimise rounding errors when calculating 2SD and 3SD windows.



			Absolute	Absolute Standard Deviations Relative Standard Deviatio								
Constituent	Certified			T		T	Ttolative		I I	370 W	indow	
	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High	
Peroxide Fus	ion ICP cont	inued								ı	T	
Ho, ppm	0.93	0.085	0.76	1.10	0.67	1.18	9.18%	18.37%	27.55%	0.88	0.97	
K, wt.%	1.82	0.051	1.72	1.93	1.67	1.98	2.82%	5.64%	8.45%	1.73	1.92	
La, ppm	43.9	2.66	38.6	49.2	36.0	51.9	6.05%	12.10%	18.14%	41.7	46.1	
Li, ppm	97	5.4	86	108	81	113	5.55%	11.10%	16.66%	92	102	
Lu, ppm	0.39	0.025	0.34	0.44	0.31	0.46	6.38%	12.76%	19.15%	0.37	0.41	
Mg, wt.%	3.36	0.066	3.23	3.49	3.16	3.56	1.95%	3.91%	5.86%	3.19	3.53	
Mn, wt.%	0.070	0.001	0.068	0.071	0.067	0.072	1.16%	2.32%	3.48%	0.066	0.073	
Nd, ppm	32.7	2.15	28.4	37.0	26.2	39.2	6.59%	13.17%	19.76%	31.1	34.3	
Ni, ppm	48.8	10.3	28.3	69.4	18.0	79.6	21.04%	42.07%	63.11%	46.4	51.3	
P, wt.%	0.052	0.004	0.044	0.059	0.041	0.063	7.20%	14.41%	21.61%	0.049	0.054	
Pb, ppm	15.7	2.3	11.2	20.3	8.9	22.6	14.49%	28.99%	43.48%	14.9	16.5	
Pr, ppm	8.85	0.384	8.08	9.62	7.70	10.00	4.34%	8.69%	13.03%	8.41	9.29	
Rb, ppm	71	2.9	65	76	62	79	4.06%	8.12%	12.18%	67	74	
S, wt.%	1.65	0.049	1.56	1.75	1.51	1.80	2.93%	5.87%	8.80%	1.57	1.74	
Sb, ppm	1.56	0.39	0.79	2.33	0.40	2.71	24.76%	49.52%	74.28%	1.48	1.63	
Sc, ppm	10.4	0.70	9.0	11.8	8.3	12.5	6.70%	13.41%	20.11%	9.9	10.9	
Si, wt.%	28.74	0.713	27.31	30.16	26.60	30.88	2.48%	4.96%	7.44%	27.30	30.18	
Sm, ppm	5.12	0.501	4.11	6.12	3.61	6.62	9.79%	19.58%	29.37%	4.86	5.37	
Sr, ppm	51	5.1	41	62	36	67	9.91%	19.82%	29.73%	49	54	
Ta, ppm	1.00	0.18	0.63	1.36	0.45	1.54	18.12%	36.23%	54.35%	0.95	1.05	
Tb, ppm	0.73	0.10	0.54	0.92	0.44	1.02	13.27%	26.54%	39.81%	0.69	0.77	
Th, ppm	9.67	1.10	7.47	11.88	6.37	12.98	11.39%	22.77%	34.16%	9.19	10.16	
Ti, wt.%	0.348	0.017	0.313	0.382	0.295	0.400	5.01%	10.02%	15.03%	0.330	0.365	
Tm, ppm	0.40	0.019	0.36	0.44	0.34	0.45	4.76%	9.51%	14.27%	0.38	0.42	
U, ppm	19.7	0.70	18.3	21.1	17.6	21.8	3.56%	7.12%	10.68%	18.7	20.7	
V, ppm	123	8	107	140	99	148	6.64%	13.28%	19.92%	117	130	
Y, ppm	25.2	1.28	22.6	27.8	21.4	29.0	5.08%	10.15%	15.23%	23.9	26.5	
Yb, ppm	2.58	0.204	2.17	2.99	1.97	3.19	7.91%	15.83%	23.74%	2.45	2.71	
Zn, ppm	42.0	8.3	25.5	58.6	17.2	66.9	19.69%	39.38%	59.07%	39.9	44.2	
*4-Acid Diges		0.0	20.0	30.0	17.2	00.5	13.0370	00.0070	00.01 70	00.0	77.2	
Ag, ppm	0.177	0.026	0.125	0.228	0.100	0.253	14.47%	28.94%	43.41%	0.168	0.185	
Al, wt.%	4.93	0.168	4.59	5.26	4.43	5.43	3.40%	6.80%	10.21%	4.68	5.18	
	28.5	2.73	23.0	34.0	20.3	36.7	9.58%	19.17%	28.75%	27.1	29.9	
As, ppm												
Ba, ppm	496	16	464	528	448	543	3.22%	6.43%	9.65%	471	520	
Be, ppm	2.78	0.149	2.48	3.08	2.33	3.23	5.37%	10.74%	16.11%	2.64	2.92	
Bi, ppm	4.39	0.299	3.79	4.98	3.49	5.28	6.82%	13.65%	20.47%	4.17	4.61	
Ca, wt.%	2.76	0.070	2.62	2.90	2.55	2.97	2.55%	5.10%	7.65%	2.62	2.90	
Cd, ppm	0.25	0.015	0.22	0.28	0.20	0.29	6.03%	12.06%	18.09%	0.24	0.26	
Ce, ppm	77	4.9	67	86	62	91	6.37%	12.74%	19.11%	73	80	
Co, wt.%	2.15	0.069	2.01	2.29	1.94	2.36	3.19%	6.37%	9.56%	2.04	2.26	
Cr, ppm	55	8	38	71	30	79	15.05%	30.10%	45.14%	52	57	
Cs, ppm	2.01	0.130	1.75	2.27	1.62	2.40	6.47%	12.94%	19.41%	1.91	2.11	

SI unit equivalents: ppm, parts per million \equiv mg/kg \equiv μ g/g \equiv 0.0001 wt.% \equiv 1000 ppb, parts per billion.

Note: intervals may appear asymmetric due to rounding; the number of decimal places quoted does not imply accuracy of the certified value to this level but are given to minimise rounding errors when calculating 2SD and 3SD windows.



^{*}Four acid digestion quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.

Absolute Standard Deviations Relative Standard Deviations 5% window												
Constituent	Certified		Absolute	Standard	Deviations	S	Relative	Standard D	eviations	5% w	indow	
Constituent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High	
*4-Acid Diges	tion continu	ued										
Cu, wt.%	2.20	0.038	2.13	2.28	2.09	2.32	1.73%	3.46%	5.19%	2.09	2.31	
Dy, ppm	3.86	0.344	3.17	4.55	2.83	4.89	8.91%	17.83%	26.74%	3.66	4.05	
Er, ppm	2.33	0.159	2.01	2.65	1.85	2.81	6.84%	13.68%	20.53%	2.21	2.45	
Eu, ppm	0.76	0.058	0.64	0.88	0.59	0.94	7.65%	15.29%	22.94%	0.72	0.80	
Fe, wt.%	0.820	0.035	0.750	0.889	0.715	0.924	4.27%	8.54%	12.80%	0.779	0.861	
Ga, ppm	13.2	0.93	11.4	15.1	10.5	16.0	7.01%	14.03%	21.04%	12.6	13.9	
Gd, ppm	4.23	0.140	3.95	4.51	3.81	4.65	3.32%	6.63%	9.95%	4.01	4.44	
Ge, ppm	0.098	0.029	0.041	0.156	0.012	0.185	29.22%	58.43%	87.65%	0.093	0.103	
Hf, ppm	3.43	0.226	2.97	3.88	2.75	4.10	6.60%	13.19%	19.79%	3.26	3.60	
Ho, ppm	0.79	0.08	0.62	0.96	0.54	1.04	10.62%	21.24%	31.86%	0.75	0.83	
In, ppm	0.14	0.011	0.12	0.16	0.11	0.18	7.74%	15.48%	23.22%	0.14	0.15	
K, wt.%	1.83	0.065	1.70	1.96	1.64	2.03	3.55%	7.11%	10.66%	1.74	1.92	
La, ppm	37.8	2.67	32.5	43.2	29.8	45.9	7.06%	14.11%	21.17%	36.0	39.7	
Li, ppm	96	5.2	86	107	81	112	5.40%	10.80%	16.21%	91	101	
Lu, ppm	0.33	0.026	0.28	0.39	0.25	0.41	7.92%	15.83%	23.75%	0.32	0.35	
Mg, wt.%	3.35	0.088	3.18	3.53	3.09	3.62	2.63%	5.25%	7.88%	3.19	3.52	
Mn, wt.%	0.066	0.003	0.060	0.072	0.057	0.075	4.61%	9.23%	13.84%	0.063	0.069	
Mo, ppm	2.24	0.140	1.96	2.52	1.82	2.66	6.26%	12.52%	18.78%	2.13	2.35	
Na, wt.%	0.053	0.006	0.041	0.065	0.034	0.072	11.73%	23.46%	35.20%	0.050	0.056	
Nb, ppm	10.5	2.1	6.2	14.7	4.0	16.9	20.52%	41.04%	61.55%	9.9	11.0	
Nd, ppm	29.4	2.90	23.6	35.2	20.7	38.1	9.86%	19.73%	29.59%	27.9	30.8	
Ni, ppm	42.4	1.92	38.6	46.3	36.7	48.2	4.52%	9.03%	13.55%	40.3	44.6	
P, wt.%	0.049	0.002	0.045	0.053	0.042	0.055	4.31%	8.62%	12.93%	0.046	0.051	
Pb, ppm	15.2	0.92	13.4	17.1	12.5	18.0	6.03%	12.05%	18.08%	14.5	16.0	
Pr, ppm	7.59	0.86	5.87	9.30	5.01	10.16	11.31%	22.61%	33.92%	7.21	7.97	
Rb, ppm	71	3.7	63	78	60	82	5.19%	10.39%	15.58%	67	74	
Re, ppm	0.041	0.002	0.036	0.045	0.034	0.048	5.80%	11.60%	17.40%	0.039	0.043	
S, wt.%	1.70	0.045	1.61	1.79	1.56	1.83	2.62%	5.24%	7.87%	1.61	1.78	
Sb, ppm	1.49	0.095	1.30	1.68	1.21	1.78	6.39%	12.79%	19.18%	1.42	1.57	
Sc, ppm	10.5	0.36	9.8	11.2	9.4	11.6	3.48%	6.96%	10.43%	10.0	11.0	
Se, ppm	3.74	0.65	2.44	5.05	1.79	5.70	17.43%	34.86%	52.29%	3.56	3.93	
Sm, ppm	4.89	0.49	3.90	5.88	3.41	6.37	10.09%	20.17%	30.26%	4.65	5.14	
Sn, ppm	1.71	0.147	1.42	2.01	1.27	2.15	8.61%	17.22%	25.83%	1.63	1.80	
Sr, ppm	46.9	2.90	41.1	52.7	38.2	55.6	6.18%	12.36%	18.53%	44.6	49.3	
Ta, ppm	0.75	0.19	0.38	1.12	0.19	1.31	24.90%	49.80%	74.71%	0.71	0.79	
Tb, ppm	0.66	0.043	0.58	0.75	0.53	0.79	6.46%	12.92%	19.37%	0.63	0.69	
Te, ppm	< 0.05	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND	
Th, ppm	9.52	0.474	8.57	10.47	8.10	10.94	4.98%	9.96%	14.95%	9.04	9.99	
Ti, wt.%	0.238	0.057	0.125	0.352	0.068	0.409	23.85%	47.70%	71.56%	0.226	0.250	
TI, ppm	0.38	0.026	0.33	0.44	0.31	0.46	6.64%	13.28%	19.92%	0.37	0.40	
Tm, ppm	0.34	0.034	0.27	0.41	0.24	0.44	9.91%	19.82%	29.73%	0.32	0.36	
U, ppm	18.8	1.17	16.4	21.1	15.3	22.3	6.22%	12.43%	18.65%	17.8	19.7	
		narte ner			l		l .			<u> </u>	l	

SI unit equivalents: ppm, parts per million \equiv mg/kg \equiv μ g/g \equiv 0.0001 wt.% \equiv 1000 ppb, parts per billion.

Note: intervals may appear asymmetric due to rounding; the number of decimal places quoted does not imply accuracy of the certified value to this level but are given to minimise rounding errors when calculating 2SD and 3SD windows.



^{*}Four acid digestion quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.

	Certified		Absolute	Standard	Deviations	3	Relative	Standard D	eviations	5% w	indow
Constituent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High
*4-Acid Diges	tion continu	ued		l rugu							
V, ppm	121	8	105	136	98	144	6.32%	12.65%	18.97%	115	127
W, ppm	1.26	0.17	0.92	1.60	0.75	1.77	13.58%	27.16%	40.74%	1.20	1.32
Y, ppm	21.4	1.03	19.4	23.5	18.3	24.5	4.81%	9.61%	14.42%	20.4	22.5
Yb, ppm	2.23	0.201	1.83	2.63	1.63	2.83	9.00%	18.01%	27.01%	2.12	2.34
Zn, ppm	35.9	3.14	29.6	42.2	26.5	45.3	8.74%	17.48%	26.22%	34.1	37.7
Zr, ppm	122	11	100	143	90	154	8.80%	17.59%	26.39%	116	128
Aqua Regia D	igestion			L						L	
Ag, ppm	0.141	0.010	0.120	0.162	0.110	0.173	7.43%	14.86%	22.29%	0.134	0.148
Al, wt.%	1.31	0.15	1.00	1.61	0.85	1.76	11.57%	23.15%	34.72%	1.24	1.37
As, ppm	25.0	2.20	20.6	29.4	18.4	31.6	8.82%	17.64%	26.47%	23.7	26.2
Ba, ppm	156	43	70	241	28	284	27.39%	54.77%	82.16%	148	163
Be, ppm	1.50	0.073	1.35	1.64	1.28	1.71	4.85%	9.70%	14.55%	1.42	1.57
Bi, ppm	3.70	0.199	3.30	4.10	3.10	4.30	5.38%	10.75%	16.13%	3.52	3.89
Ca, wt.%	2.69	0.108	2.47	2.91	2.37	3.01	4.01%	8.03%	12.04%	2.56	2.82
Cd, ppm	0.25	0.019	0.21	0.28	0.19	0.30	7.80%	15.59%	23.39%	0.23	0.26
Ce, ppm	26.5	2.15	22.2	30.8	20.1	33.0	8.11%	16.22%	24.34%	25.2	27.9
Co, wt.%	1.99	0.155	1.68	2.30	1.52	2.45	7.80%	15.61%	23.41%	1.89	2.09
Cr, ppm	21.2	2.9	15.4	27.0	12.5	29.9	13.65%	27.31%	40.96%	20.2	22.3
Cs, ppm	0.26	0.08	0.10	0.41	0.02	0.49	30.28%	60.55%	90.83%	0.24	0.27
Cu, wt.%	2.12	0.092	1.93	2.30	1.84	2.39	4.32%	8.64%	12.97%	2.01	2.22
Dy, ppm	1.73	0.113	1.51	1.96	1.39	2.07	6.53%	13.06%	19.59%	1.65	1.82
Er, ppm	0.80	0.040	0.72	0.88	0.68	0.92	5.00%	10.00%	15.00%	0.76	0.84
Eu, ppm	0.40	0.029	0.35	0.46	0.32	0.49	7.16%	14.33%	21.49%	0.38	0.42
Fe, wt.%	0.672	0.025	0.622	0.723	0.596	0.749	3.78%	7.56%	11.34%	0.639	0.706
Ga, ppm	3.46	0.46	2.55	4.37	2.09	4.83	13.22%	26.43%	39.65%	3.29	3.63
Gd, ppm	2.33	0.224	1.88	2.78	1.66	3.00	9.63%	19.26%	28.88%	2.21	2.44
Hf, ppm	0.41	0.06	0.30	0.53	0.24	0.58	13.66%	27.31%	40.97%	0.39	0.43
Hg, ppm	0.048	0.010	0.028	0.068	0.018	0.078	20.72%	41.45%	62.17%	0.046	0.051
Ho, ppm	0.30	0.009	0.28	0.32	0.27	0.33	3.01%	6.01%	9.02%	0.28	0.31
In, ppm	0.092	0.006	0.080	0.103	0.074	0.109	6.45%	12.89%	19.34%	0.087	0.096
K, wt.%	0.276	0.029	0.218	0.334	0.189	0.363	10.50%	20.99%	31.49%	0.263	0.290
La, ppm	7.41	1.46	4.50	10.32	3.04	11.78	19.66%	39.31%	58.97%	7.04	7.78
Li, ppm	61	9	43	79	34	88	14.70%	29.40%	44.10%	58	64
Lu, ppm	0.095	0.005	0.085	0.105	0.080	0.110	5.31%	10.61%	15.92%	0.090	0.099
Mg, wt.%	2.82	0.112	2.59	3.04	2.48	3.16	3.98%	7.95%	11.93%	2.68	2.96
Mn, wt.%	0.064	0.003	0.058	0.070	0.055	0.073	4.84%	9.68%	14.52%	0.061	0.067
Mo, ppm	1.98	0.20	1.58	2.37	1.38	2.57	10.09%	20.18%	30.28%	1.88	2.07
Nd, ppm	10.1	0.85	8.4	11.8	7.6	12.6	8.39%	16.77%	25.16%	9.6	10.6
Ni, ppm	36.2	4.1	28.0	44.4	23.9	48.5	11.32%	22.64%	33.96%	34.4	38.0
P, wt.%	0.042	0.002	0.038	0.046	0.036	0.048	4.85%	9.69%	14.54%	0.040	0.044
Pb, ppm	12.5	0.83	10.8	14.1	10.0	14.9	6.66%	13.32%	19.98%	11.8	13.1
Pr, ppm	2.14	0.39	1.35	2.93	0.96	3.33	18.41%	36.83%	55.24%	2.04	2.25
SI unit equival										•	•

SI unit equivalents: ppm, parts per million \equiv mg/kg \equiv μ g/g \equiv 0.0001 wt.% \equiv 1000 ppb, parts per billion.

Note: intervals may appear asymmetric due to rounding; the number of decimal places quoted does not imply accuracy of the certified value to this level but are given to minimise rounding errors when calculating 2SD and 3SD windows.



^{*}Four acid digestion quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.

0	Certified		Absolute	Standard	Deviations	3	Relative	Standard D	eviations	5% w	indow
Constituent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High
Aqua Regia D	igestion co	ntinued									
Rb, ppm	9.00	1.11	6.79	11.22	5.68	12.33	12.31%	24.63%	36.94%	8.55	9.45
Re, ppm	0.039	0.003	0.034	0.045	0.031	0.047	6.90%	13.80%	20.70%	0.037	0.041
S, wt.%	1.41	0.23	0.95	1.86	0.73	2.09	16.15%	32.30%	48.45%	1.34	1.48
Sb, ppm	0.91	0.15	0.60	1.21	0.45	1.36	16.76%	33.52%	50.27%	0.86	0.95
Sc, ppm	5.12	0.334	4.45	5.79	4.12	6.12	6.53%	13.05%	19.58%	4.86	5.37
Se, ppm	3.38	0.39	2.60	4.15	2.21	4.54	11.50%	23.00%	34.51%	3.21	3.55
Sn, ppm	0.45	0.06	0.33	0.57	0.28	0.63	12.94%	25.88%	38.82%	0.43	0.47
Sr, ppm	23.4	1.43	20.5	26.2	19.1	27.7	6.13%	12.25%	18.38%	22.2	24.5
Ta, ppm	< 0.01	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND
Tb, ppm	0.33	0.016	0.30	0.36	0.28	0.38	4.83%	9.66%	14.49%	0.32	0.35
Th, ppm	3.38	0.233	2.91	3.85	2.68	4.08	6.90%	13.80%	20.70%	3.21	3.55
Ti, wt.%	< 0.005	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND
TI, ppm	0.17	0.016	0.14	0.20	0.12	0.22	9.38%	18.75%	28.13%	0.16	0.18
U, ppm	14.4	0.94	12.5	16.3	11.6	17.2	6.49%	12.98%	19.48%	13.7	15.1
V, ppm	40.2	2.19	35.8	44.5	33.6	46.7	5.44%	10.89%	16.33%	38.2	42.2
W, ppm	0.22	0.03	0.17	0.28	0.14	0.30	11.74%	23.48%	35.22%	0.21	0.23
Y, ppm	7.91	0.358	7.20	8.63	6.84	8.99	4.53%	9.05%	13.58%	7.52	8.31
Yb, ppm	0.67	0.027	0.62	0.73	0.59	0.76	4.07%	8.14%	12.20%	0.64	0.71
Zn, ppm	30.9	2.38	26.2	35.7	23.8	38.0	7.68%	15.37%	23.05%	29.4	32.5
Zr, ppm	13.6	1.11	11.3	15.8	10.2	16.9	8.20%	16.41%	24.61%	12.9	14.2

SI unit equivalents: ppm, parts per million ≡ mg/kg ≡ μg/g ≡ 0.0001 wt.% ≡ 1000 ppb, parts per billion.

Note: intervals may appear asymmetric due to rounding; the number of decimal places quoted does not imply accuracy of the certified value to this level but are given to minimise rounding errors when calculating 2SD and 3SD windows.

Table 2. Indicative Values for OREAS 556.

Constituent	Unit	Value	Constituent	Unit	Value	Constituent	Unit	Value
Borate Fusio	on XRF							
As	ppm	< 10	Na₂O	wt.%	0.073	V ₂ O ₅	ppm	240
Bi	ppm	< 10	Ni	ppm	220	Zn	ppm	43.9
Cd	ppm	< 10	Pb	ppm	< 10	Zr	ppm	138
CI	ppm	350	Sb	ppm	< 10			
Мо	ppm	21.0	Sn	ppm	< 10			
Thermograv	imetry							
H ₂ O-	wt.%	0.060						
Sulphuric A	cid 10%	Leach						
Cu	wt.%	0.951						
Peroxide Fu	sion ICP							
Be	ppm	2.86	In	ppm	0.15	Sn	ppm	1.64
Cd	ppm	0.25	Мо	ppm	2.75	Te	ppm	< 1
Ge	ppm	1.66	Nb	ppm	14.8	TI	ppm	< 0.5
Hf	ppm	4.68	Re	ppm	< 0.1	W	ppm	1.20
Hg	ppm	< 0.1	Se	ppm	10.4	Zr	ppm	161

SI unit equivalents: ppm (parts per million) \equiv mg/kg \equiv μ g/g \equiv 0.0001 wt.% \equiv 1000 ppb (parts per billion).

Note: the number of significant figures reported is not a reflection of the level of certainty of stated values. They are instead an artefact of ORE's in-house CRM-specific LIMS.

Constituent	Unit	Value	Constituent	Unit	Value	Constituent	Unit	Value
4-Acid Diges	stion							
В	ppm	11.7	Hg	ppm	0.057			
Aqua Regia	Digestio	n						
Au	ppm	0.007	Nb	ppm	< 0.05	Sm	ppm	2.32
В	ppm	15.2	Pd	ppb	< 10	Те	ppm	0.032
Ge	ppm	0.038	Pt	ppb	15.0	Tm	ppm	0.10
Na	wt.%	0.011	Si	wt.%	0.070			

SI unit equivalents: ppm (parts per million) \equiv mg/kg \equiv μ g/g \equiv 0.0001 wt.% \equiv 1000 ppb (parts per billion).

Note: the number of significant figures reported is not a reflection of the level of certainty of stated values. They are instead an artefact of ORE's in-house CRM-specific LIMS.

INTRODUCTION

OREAS reference materials are intended to provide a low-cost method of evaluating and improving the quality of analysis of geological samples. To the geologist they provide a means of implementing quality control in analytical data sets generated in exploration from the grass roots level through to prospect evaluation, and in grade control at mining operations. To the analyst they provide an effective means of calibrating analytical equipment, assessing new techniques and routinely monitoring in-house procedures. OREAS reference materials enable users to successfully achieve process control of these tasks because the observed variance from repeated analysis has its origin almost exclusively in the analytical process rather than the reference material itself. In evaluating laboratory performance with this CRM, the section headed 'Instructions for correct use' should be read carefully.

Table 1 provides performance gate intervals for the certified values. Table 2 shows indicative values, Table 3 provides some indicative physical properties and Table 4 presents the 95% confidence and tolerance limits for all certified values. Tabulated results of all elements together with uncorrected means, medians, standard deviations, relative standard deviations and per cent deviation of lab means from the corrected mean of means (PDM³) are presented in the detailed certification data for this CRM (OREAS 556-DataPack.1.1.210420_174449.xlsx).

Results are also presented in scatter plots for Co and Cu by 4-acid digestion and peroxide fusion ICP in Figures 1 to 4 respectively, together with ±3SD (magenta) and ±5% (yellow) control lines and certified value (green line). Accepted individual results are coloured blue and individual and dataset outliers are identified in red and violet, respectively.

SOURCE MATERIAL

OREAS 556 was prepared from a blend of copper-cobalt primary ore materials supplied by Metorex (subsidiary of Jinchuan Group) from the Dilala Est Copper Cobalt Project, located within the Kolwezi Klippe area of Lualaba Province (Democratic Republic of the Congo). Kolwezi is ~320 kilometres northwest from Lubumbashi, the provincial capital of Haut Katanga Province. Copper-cobalt ores within the Kolwezi Klippe area are stratiform, sediment-hosted deposits hosted within Neoproterozoic sedimentary rocks of the Central African Copperbelt. Primary mineralisation typically consists of bornite, chalcopyrite and

cobalt sulphides (linnaeite and carrollite) hosted within dolomitic argillite, shales, sandstone and dolomite horizons.

PERFORMANCE GATES

Table 1 above shows intervals calculated for two and three standard deviations. As a guide these intervals may be regarded as warning or rejection for multiple 2SD outliers, or rejection for individual 3SD outliers in QC monitoring, although their precise application should be at the discretion of the QC manager concerned (also see 'Intended Use' section below). Westgard Rules extend the basics of single-rule QC monitoring using multi-rules (for more information visit www.westgard.com/mltirule.htm). A second method utilises a 5% window calculated directly from the certified value.

Standard deviation is also shown in relative percent for one, two and three relative standard deviations (1RSD, 2RSD and 3RSD) to facilitate an appreciation of the magnitude of these numbers and a comparison with the 5% window. Caution should be exercised when concentration levels approach lower limits of detection of the analytical methods employed as performance gates calculated from standard deviations tend to be excessively wide whereas those determined by the 5% method are too narrow. One approach used at commercial laboratories is to set the acceptance criteria at twice the detection level (DL) \pm 10%.

i.e. Certified Value ± 10% ± 2DL (adapted from Govett, 1983).

COMMINUTION AND HOMOGENISATION PROCEDURES

The material constituting OREAS 556 was prepared in the following manner:

- Drying to constant mass at 105° Celsius;
- Multi-stage milling of each source material to achieve a particle size distribution of >98% passing 75 microns;
- Preliminary homogenisation of each source material;
- Representative sampling and check assaying of each source material;
- Blending in appropriate proportions to achieve target grades;
- Packaging in 10g units sealed in laminated foil pouches and 500g units in plastic jars.

PHYSICAL PROPERTIES

OREAS 556 was tested at ORE Research & Exploration Pty Ltd's onsite facility for various physical properties. Table 3 presents these findings that should be used for informational purposes only.

Table 3. Physical properties of OREAS 556.

Bulk Density (g/L)	Moisture%	Munsell Notation [‡]	Munsell Color‡
792.7	0.44	N6	Medium Light Gray

[‡]The Munsell Rock Color Chart helps geologists and archeologists communicate with colour more effectively by cross-referencing ISCC-NBS colour names with unique Munsell alpha-numeric colour notations for rock colour samples.

ANALYTICAL PROGRAM

Thirty analytical laboratories participated in the program to characterise the elements reported in Table 1. The following methods were employed:

- Lithium borate fusion whole rock analysis package by X-ray fluorescence (up to 19 laboratories depending on the element);
- Thermogravimetry: Loss on Ignition (LOI) at 1000°C (13 laboratories used a thermogravimetric analyser, 7 laboratories included LOI with their fusion package and 4 laboratories used conventional muffle furnace);
- C and S by infrared combustion furnace/CS analyser (27 laboratories);
- Cu and Co by *5% sulphuric acid leach with ICP or AAS finish (up to 19 laboratories);
- Cu and Co by *10% sulphuric acid leach with ICP or AAS finish (up to 12 laboratories);
- Sodium peroxide fusion with full suite elemental package by ICP-OES and/or MS finish (up to 20 laboratories depending on the element);
- 4-acid digestion for full suite elemental package by ICP-OES and MS finish (up to 26 laboratories depending on the element);
- Aqua regia digestion for full elemental suite ICP-OES and ICP-MS (up to 26 laboratories depending on the element).

For the round robin program ten 400g test units were taken at predetermined intervals immediately following homogenisation and are considered representative of the entire prepared batch. The six samples received by each laboratory were obtained by taking two 25g scoop splits from each of three separate 400g test units. This format enabled nested ANOVA treatment of the results to evaluate homogeneity, i.e., to ascertain whether between-unit variance is greater than within-unit variance (see 'Homogeneity Evaluation' section below).

STATISTICAL ANALYSIS

Standard Deviation intervals (see Table 1) provide an indication of a level of performance that might reasonably be expected from a laboratory being monitored by this CRM in a QA/QC program. They take into account errors attributable to measurement uncertainty and CRM variability. For an effective CRM the contribution of the latter should be negligible in comparison to measurement errors. The Standard Deviation values include all sources of measurement uncertainty: between-laboratory variance, within-run variance (precision errors) and CRM variability.

In the application of SD's in monitoring performance it is important to note that not all laboratories function at the same level of proficiency and that different methods in use at a particular laboratory have differing levels of precision. Each laboratory has its own inherent SD (for a specific concentration level and analyte-method pair) based on the analytical process and this SD is not directly related to the round robin program (see Intended Use section for more detail).

The SD for each analyte's certified value is calculated from the same filtered data set used to determine the certified value, i.e., after removal of all individual, lab dataset (batch) and

^{*}See 'Appendix' for specified methodology.

3SD outliers (single iteration). These outliers can only be removed after the absolute homogeneity of the CRM has been independently established, i.e., the outliers must be confidently deemed to be analytical rather than arising from inhomogeneity of the CRM. The standard deviation is then calculated for each analyte from the pooled accepted analyses generated from the certification program.

Certified Values, Confidence Limits and Tolerance Limits (Table 4) have been determined for each analyte following removal of individual, laboratory dataset (batch) and 3SD outliers (single iteration).

95% Confidence Limits are inversely proportional to the number of participating laboratories and inter-laboratory agreement. It is a measure of the reliability of the certified value. A 95% confidence interval indicates a 95% probability that the true value of the analyte under consideration lies between the upper and lower limits. **95% Confidence Limits should not be used as control limits for laboratory performance.**

Indicative (uncertified) values (Table 2) are provided where i) a laboratory reported analytes beyond those requested; ii) the number of laboratories reporting a particular analyte is insufficient (< 5) to support certification; iii) inter-laboratory consensus is poor; or iv) a significant proportion of results are outlying.

For individual outliers within a laboratory batch the z-score test is used in combination with a second method that determines the per cent deviation of the individual value from the batch median. Outliers in general are selected on the basis of z-scores > 2.5 and with per cent deviations (i) > 3 and (ii) more than three times the average absolute per cent deviation for the batch. In certain instances, statistician's prerogative has been employed in discriminating outliers. Each laboratory data set mean is tested for outlying status based on z-score discrimination and rejected if > 2.5. After individual and laboratory data set (batch) outliers have been eliminated a non-iterative 3 standard deviation filter is applied, with those values lying outside this window also relegated to outlying status.

Table 4. 95% Confidence & Tolerance Limits for OREAS 556.

Constituent	Certified	95% Confid	lence Limits	95% Toler	ance Limits
Constituent	Value	Low	High	Low	High
Borate Fusion XRF					
Al ₂ O ₃ , Aluminium(III) oxide (wt.%)	9.43	9.37	9.49	9.37	9.50
BaO, Barium oxide (ppm)	576	543	608	503	648
CaO, Calcium oxide (wt.%)	3.86	3.82	3.90	3.83	3.89
Co, Cobalt (wt.%)	2.10	2.07	2.13	2.08	2.12
Cr ₂ O ₃ , Chromium(III) oxide (ppm)	106	99	113	IND	IND
Cu, Copper (wt.%)	2.18	2.13	2.23	2.15	2.21
Fe ₂ O ₃ , Iron(III) oxide (wt.%)	1.18	1.16	1.19	1.16	1.19
K ₂ O, Potassium oxide (wt.%)	2.15	2.13	2.17	2.13	2.17
MgO, Magnesium oxide (wt.%)	5.58	5.55	5.61	5.54	5.63
MnO, Manganese oxide (wt.%)	0.089	0.086	0.092	0.087	0.090
P ₂ O ₅ , Phosphorus(V) oxide (wt.%)	0.113	0.110	0.116	0.111	0.115
SiO ₂ , Silicon dioxide (wt.%)	60.82	60.50	61.13	60.51	61.12

SI unit equivalents: ppm, parts per million ≡ mg/kg ≡ μg/g ≡ 0.0001 wt.% ≡ 1000 ppb, parts per billion.



	Table 1 contin	1				
Constituent	Certified	95% Confid	lence Limits			
Constituent	Value	Low	High	Low	High	
Borate Fusion XRF continued						
SO ₃ , Sulphur trioxide (wt.%)	4.22	4.19	4.26	4.17	4.28	
SrO, Strontium oxide (ppm)	84	68	100	IND	IND	
TiO ₂ , Titanium dioxide (wt.%)	0.577	0.570	0.583	0.564	0.589	
Thermogravimetry						
LOI ¹⁰⁰⁰ , Loss On Ignition @1000°C (wt.%)	7.27	7.01	7.54	7.18	7.37	
Infrared Combustion						
C, Carbon (wt.%)	1.80	1.79	1.81	1.78	1.81	
S, Sulphur (wt.%)	1.69	1.67	1.70	1.66	1.71	
Sulphuric Acid 5% Leach						
Co, Cobalt (wt.%)	0.823	0.738	0.907	0.800	0.845	
Sulphuric Acid 10% Leach						
Co, Cobalt (wt.%)	0.828	0.780	0.877	0.809	0.848	
Peroxide Fusion ICP			1		•	
Ag, Silver (ppm)	< 1	IND	IND	IND	IND	
Al, Aluminium (wt.%)	4.95	4.89	5.01	4.86	5.04	
As, Arsenic (ppm)	30.0	27.1	33.0	27.8	32.3	
B, Boron (ppm)	475	445	505	462	488	
Ba, Barium (ppm)	501	489	513	486	516	
Bi, Bismuth (ppm)	4.66	4.25	5.06	4.23	5.08	
Ca, Calcium (wt.%)	2.80	2.76	2.84	2.74	2.86	
Ce, Cerium (ppm)	83	82	85	81	86	
Co, Cobalt (wt.%)	2.13	2.12	2.14	2.08	2.18	
Cr, Chromium (ppm)	78	71	86	68	89	
Cs, Caesium (ppm)	2.09	1.95	2.22	1.93	2.24	
Cu, Copper (wt.%)	2.20	2.17	2.23	2.15	2.25	
Dy, Dysprosium (ppm)	4.51	4.31	4.70	4.19	4.82	
Er, Erbium (ppm)	2.74	2.61	2.86	2.57	2.90	
Eu, Europium (ppm)	0.80	0.73	0.86	0.75	0.84	
Fe, Iron (wt.%)	0.836	0.821	0.851	0.816	0.856	
Ga, Gallium (ppm)	14.0	13.1	15.0	13.2	14.9	
Gd, Gadolinium (ppm)	4.85	4.26	5.43	4.67	5.03	
Ho, Holmium (ppm)	0.93	0.87	0.98	0.87	0.98	
K, Potassium (wt.%)	1.82	1.80	1.85	1.78	1.86	
La, Lanthanum (ppm)	43.9	42.2	45.7	42.2	45.7	
Li, Lithium (ppm)	97	93	101	94	101	
Lu, Lutetium (ppm)	0.39	0.37	0.41	0.36	0.41	
Mg, Magnesium (wt.%)	3.36	3.33	3.39	3.31	3.41	
Mn, Manganese (wt.%)	0.070	0.069	0.070	0.068	0.071	
	•			•		

SI unit equivalents: ppm, parts per million \equiv mg/kg \equiv μ g/g \equiv 0.0001 wt.% \equiv 1000 ppb, parts per billion.



Table 4 continued. Certified 95% Confidence Limits 95% Tolerance Limits									
Constituent	Certified								
Peroxide Fusion ICP	Value	Low	High	Low	High				
	32.7	31.2	34.2	31.3	34.1				
Nd, Neodymium (ppm)									
Ni, Nickel (ppm)	48.8	38.8	58.8	44.7	52.9				
P, Phosphorus (wt.%)	0.052	0.048	0.055	IND	IND				
Pb, Lead (ppm)	15.7	13.8	17.7	14.6	16.9				
Pr, Praseodymium (ppm)	8.85	8.64	9.07	8.44	9.26				
Rb, Rubidium (ppm)	71	69	72	69	73				
S, Sulphur (wt.%)	1.65	1.62	1.68	1.62	1.69				
Sb, Antimony (ppm)	1.56	1.04	2.07	IND	IND				
Sc, Scandium (ppm)	10.4	9.8	11.0	IND	IND				
Si, Silicon (wt.%)	28.74	28.35	29.13	28.20	29.28				
Sm, Samarium (ppm)	5.12	4.81	5.42	4.79	5.44				
Sr, Strontium (ppm)	51	49	54	49	54				
Ta, Tantalum (ppm)	1.00	0.87	1.13	IND	IND				
Tb, Terbium (ppm)	0.73	0.67	0.79	0.70	0.76				
Th, Thorium (ppm)	9.67	9.02	10.33	9.20	10.15				
Ti, Titanium (wt.%)	0.348	0.340	0.355	0.339	0.356				
Tm, Thulium (ppm)	0.40	0.39	0.41	0.36	0.44				
U, Uranium (ppm)	19.7	19.2	20.1	18.8	20.6				
V, Vanadium (ppm)	123	118	129	118	129				
Y, Yttrium (ppm)	25.2	24.5	25.9	24.1	26.3				
Yb, Ytterbium (ppm)	2.58	2.45	2.70	2.39	2.77				
Zn, Zinc (ppm)	42.0	36.9	47.2	34.4	49.7				
*4-Acid Digestion									
Ag, Silver (ppm)	0.177	0.161	0.192	IND	IND				
Al, Aluminium (wt.%)	4.93	4.86	5.00	4.86	5.00				
As, Arsenic (ppm)	28.5	27.3	29.7	27.6	29.4				
Ba, Barium (ppm)	496	489	503	483	508				
Be, Beryllium (ppm)	2.78	2.72	2.85	2.68	2.89				
Bi, Bismuth (ppm)	4.39	4.25	4.52	4.27	4.50				
Ca, Calcium (wt.%)	2.76	2.74	2.79	2.72	2.81				
Cd, Cadmium (ppm)	0.25	0.24	0.25	0.23	0.27				
Ce, Cerium (ppm)	77	74	79	75	79				
Co, Cobalt (wt.%)	2.15	2.12	2.18	2.11	2.19				
Cr, Chromium (ppm)	55	51	58	51	58				
Cs, Caesium (ppm)	2.01	1.95	2.07	1.95	2.08				
Cu, Copper (wt.%)	2.20	2.19	2.21	2.17	2.23				
Dy, Dysprosium (ppm)	3.86	3.63	4.09	3.74	3.98				
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SI unit equivalents: ppm, parts per million \equiv mg/kg \equiv μ g/g \equiv 0.0001 wt.% \equiv 1000 ppb, parts per billion.



^{*}Four acid digestion quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.

2	Certified	95% Confidence Limits		95% Tolerance Limits	
Constituent	Value	Low	High	Low	High
*4-Acid Digestion continued	1				
Er, Erbium (ppm)	2.33	2.22	2.44	2.24	2.42
Eu, Europium (ppm)	0.76	0.74	0.78	0.70	0.83
Fe, Iron (wt.%)	0.820	0.804	0.835	0.803	0.836
Ga, Gallium (ppm)	13.2	12.8	13.7	12.9	13.6
Gd, Gadolinium (ppm)	4.23	4.14	4.32	4.02	4.43
Ge, Germanium (ppm)	0.098	0.073	0.123	IND	IND
Hf, Hafnium (ppm)	3.43	3.34	3.52	3.30	3.55
Ho, Holmium (ppm)	0.79	0.73	0.85	0.76	0.82
In, Indium (ppm)	0.14	0.14	0.15	0.13	0.15
K, Potassium (wt.%)	1.83	1.81	1.86	1.79	1.87
La, Lanthanum (ppm)	37.8	36.5	39.2	36.9	38.8
Li, Lithium (ppm)	96	94	99	94	99
Lu, Lutetium (ppm)	0.33	0.31	0.36	0.31	0.36
Mg, Magnesium (wt.%)	3.35	3.31	3.39	3.30	3.40
Mn, Manganese (wt.%)	0.066	0.065	0.067	0.065	0.067
Mo, Molybdenum (ppm)	2.24	2.19	2.28	2.12	2.35
Na, Sodium (wt.%)	0.053	0.050	0.056	0.051	0.055
Nb, Niobium (ppm)	10.5	9.4	11.5	10.0	10.9
Nd, Neodymium (ppm)	29.4	27.5	31.2	28.3	30.4
Ni, Nickel (ppm)	42.4	41.7	43.2	41.1	43.8
P, Phosphorus (wt.%)	0.049	0.048	0.050	0.047	0.051
Pb, Lead (ppm)	15.2	14.8	15.7	14.8	15.7
Pr, Praseodymium (ppm)	7.59	7.01	8.16	7.37	7.80
Rb, Rubidium (ppm)	71	69	72	69	72
Re, Rhenium (ppm)	0.041	0.040	0.041	0.037	0.044
S, Sulphur (wt.%)	1.70	1.68	1.72	1.67	1.73
Sb, Antimony (ppm)	1.49	1.44	1.54	1.43	1.55
Sc, Scandium (ppm)	10.5	10.4	10.6	10.1	10.9
Se, Selenium (ppm)	3.74	3.41	4.07	3.44	4.05
Sm, Samarium (ppm)	4.89	4.58	5.20	4.70	5.09
Sn, Tin (ppm)	1.71	1.65	1.78	1.63	1.79
Sr, Strontium (ppm)	46.9	45.7	48.2	45.8	48.0
Ta, Tantalum (ppm)	0.75	0.65	0.85	0.72	0.78
Tb, Terbium (ppm)	0.66	0.63	0.69	0.62	0.70
Te, Tellurium (ppm)	< 0.05	IND	IND	IND	IND
Th, Thorium (ppm)	9.52	9.31	9.73	9.30	9.74
Ti, Titanium (wt.%)	0.238	0.213	0.264	0.231	0.246

SI unit equivalents: ppm, parts per million \equiv mg/kg \equiv μ g/g \equiv 0.0001 wt.% \equiv 1000 ppb, parts per billion.



^{*}Four acid digestion quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.

rable 4 continued.						
Constituent	Certified	95% Confid	95% Confidence Limits		95% Tolerance Limits	
	Value	Low	High	Low	High	
*4-Acid Digestion continued		1				
TI, Thallium (ppm)	0.38	0.37	0.40	0.37	0.40	
Tm, Thulium (ppm)	0.34	0.31	0.37	0.32	0.36	
U, Uranium (ppm)	18.8	18.2	19.3	18.3	19.3	
V, Vanadium (ppm)	121	117	124	118	124	
W, Tungsten (ppm)	1.26	1.19	1.32	IND	IND	
Y, Yttrium (ppm)	21.4	21.0	21.9	20.8	22.0	
Yb, Ytterbium (ppm)	2.23	2.10	2.36	2.16	2.31	
Zn, Zinc (ppm)	35.9	34.6	37.3	34.6	37.3	
Zr, Zirconium (ppm)	122	117	127	119	125	
Aqua Regia Digestion	·					
Ag, Silver (ppm)	0.141	0.137	0.146	0.117	0.166	
Al, Aluminium (wt.%)	1.31	1.24	1.38	1.26	1.35	
As, Arsenic (ppm)	25.0	24.0	26.0	24.2	25.8	
Ba, Barium (ppm)	156	131	181	150	161	
Be, Beryllium (ppm)	1.50	1.46	1.53	1.44	1.55	
Bi, Bismuth (ppm)	3.70	3.61	3.79	3.60	3.80	
Ca, Calcium (wt.%)	2.69	2.65	2.73	2.63	2.75	
Cd, Cadmium (ppm)	0.25	0.24	0.25	0.22	0.27	
Ce, Cerium (ppm)	26.5	25.4	27.6	25.7	27.4	
Co, Cobalt (wt.%)	1.99	1.88	2.09	1.95	2.03	
Cr, Chromium (ppm)	21.2	19.9	22.6	20.1	22.4	
Cs, Caesium (ppm)	0.26	0.21	0.31	0.24	0.27	
Cu, Copper (wt.%)	2.12	2.07	2.16	2.08	2.16	
Dy, Dysprosium (ppm)	1.73	1.59	1.88	IND	IND	
Er, Erbium (ppm)	0.80	0.75	0.84	IND	IND	
Eu, Europium (ppm)	0.40	0.37	0.44	IND	IND	
Fe, Iron (wt.%)	0.672	0.660	0.685	0.660	0.685	
Ga, Gallium (ppm)	3.46	3.21	3.71	3.34	3.58	
Gd, Gadolinium (ppm)	2.33	2.02	2.64	2.21	2.45	
Hf, Hafnium (ppm)	0.41	0.38	0.45	0.39	0.44	
Hg, Mercury (ppm)	0.048	0.043	0.053	0.040	0.056	
Ho, Holmium (ppm)	0.30	0.29	0.31	IND	IND	
In, Indium (ppm)	0.092	0.089	0.094	0.087	0.096	
K, Potassium (wt.%)	0.276	0.263	0.290	0.267	0.286	
La, Lanthanum (ppm)	7.41	6.74	8.08	7.12	7.69	
Li, Lithium (ppm)	61	56	66	59	63	
Lu, Lutetium (ppm)	0.095	0.089	0.100	IND	IND	
					I.	

SI unit equivalents: ppm, parts per million \equiv mg/kg \equiv μ g/g \equiv 0.0001 wt.% \equiv 1000 ppb, parts per billion.



^{*}Four acid digestion quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.

Constituent	Certified	95% Confidence Limits		95% Tolerance Limits	
	Value	Low	High	Low	High
Aqua Regia Digestion continued					
Mg, Magnesium (wt.%)	2.82	2.76	2.88	2.77	2.87
Mn, Manganese (wt.%)	0.064	0.063	0.065	0.063	0.065
Mo, Molybdenum (ppm)	1.98	1.89	2.06	1.89	2.06
Nd, Neodymium (ppm)	10.1	9.0	11.2	9.7	10.5
Ni, Nickel (ppm)	36.2	34.4	38.0	35.3	37.1
P, Phosphorus (wt.%)	0.042	0.041	0.043	0.041	0.043
Pb, Lead (ppm)	12.5	12.1	12.9	12.2	12.8
Pr, Praseodymium (ppm)	2.14	1.61	2.67	2.01	2.27
Rb, Rubidium (ppm)	9.00	8.39	9.61	8.72	9.29
Re, Rhenium (ppm)	0.039	0.038	0.041	0.037	0.041
S, Sulphur (wt.%)	1.41	1.29	1.52	1.37	1.44
Sb, Antimony (ppm)	0.91	0.83	0.99	0.85	0.97
Sc, Scandium (ppm)	5.12	4.95	5.28	4.95	5.29
Se, Selenium (ppm)	3.38	3.21	3.54	2.97	3.78
Sn, Tin (ppm)	0.45	0.42	0.48	0.42	0.49
Sr, Strontium (ppm)	23.4	22.7	24.0	22.8	23.9
Ta, Tantalum (ppm)	< 0.01	IND	IND	IND	IND
Tb, Terbium (ppm)	0.33	0.32	0.35	0.31	0.35
Th, Thorium (ppm)	3.38	3.29	3.47	3.24	3.52
Ti, Titanium (wt.%)	< 0.005	IND	IND	IND	IND
TI, Thallium (ppm)	0.17	0.17	0.18	0.16	0.18
U, Uranium (ppm)	14.4	14.0	14.9	14.0	14.8
V, Vanadium (ppm)	40.2	39.1	41.3	38.8	41.5
W, Tungsten (ppm)	0.22	0.21	0.24	0.20	0.24
Y, Yttrium (ppm)	7.91	7.73	8.09	7.74	8.08
Yb, Ytterbium (ppm)	0.67	0.66	0.69	IND	IND
Zn, Zinc (ppm)	30.9	29.8	32.0	29.8	32.1
Zr, Zirconium (ppm)	13.6	13.0	14.1	13.2	13.9

SI unit equivalents: ppm, parts per million ≡ mg/kg ≡ μg/g ≡ 0.0001 wt.% ≡ 1000 ppb, parts per billion.

Note: intervals may appear asymmetric due to rounding; IND = indeterminate (due to reading resolution error).

Homogeneity Evaluation

The tolerance limits (ISO 16269:2014) shown in Table 4 were determined using an analysis of precision errors method and are considered a conservative estimate of true homogeneity. The meaning of tolerance limits may be illustrated for copper by fusion with XRF, where 99% of the time (1- α =0.99) at least 95% of subsamples (ρ =0.95) will have concentrations lying between 2.15 and 2.21 wt.%. Put more precisely, this means that if the same number of subsamples were taken and analysed in the same manner repeatedly, 99% of the tolerance intervals so constructed would cover at least 95% of the total population, and 1% of the tolerance intervals would cover less than 95% of the total population (ISO Guide 35). *Please note that tolerance limits pertain to the*

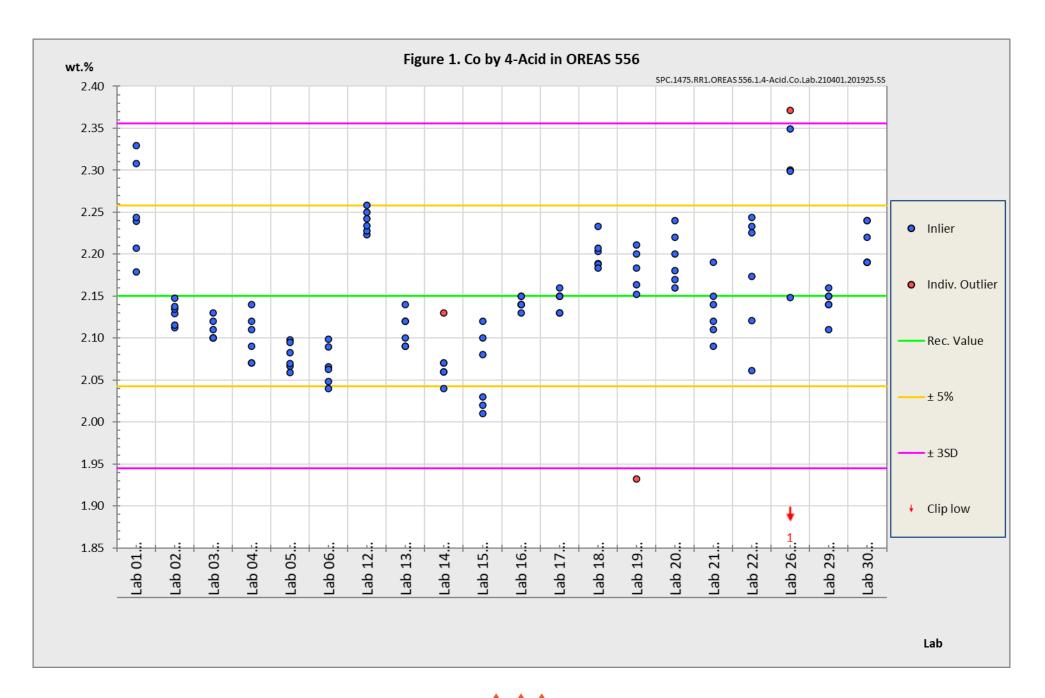
homogeneity of the CRM only and should not be used as control limits for laboratory performance.

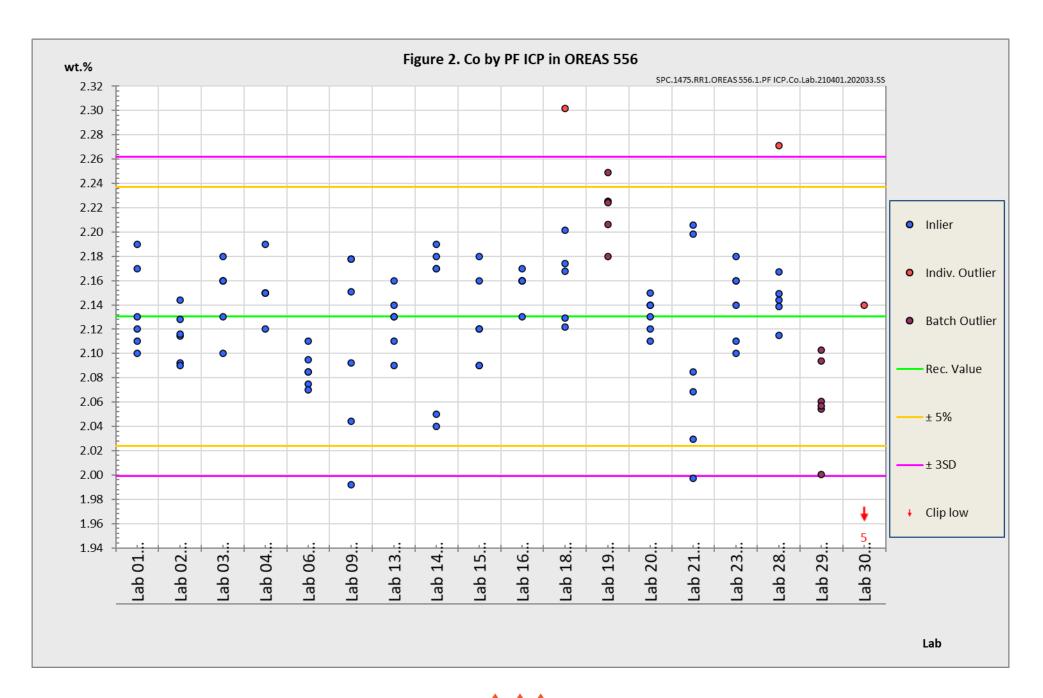
Based on the statistical analysis of the results of the inter-laboratory certification program it can be concluded that OREAS 556 is sufficiently homogenous and fit-for-purpose as a certified reference material (see 'Intended Use' below).

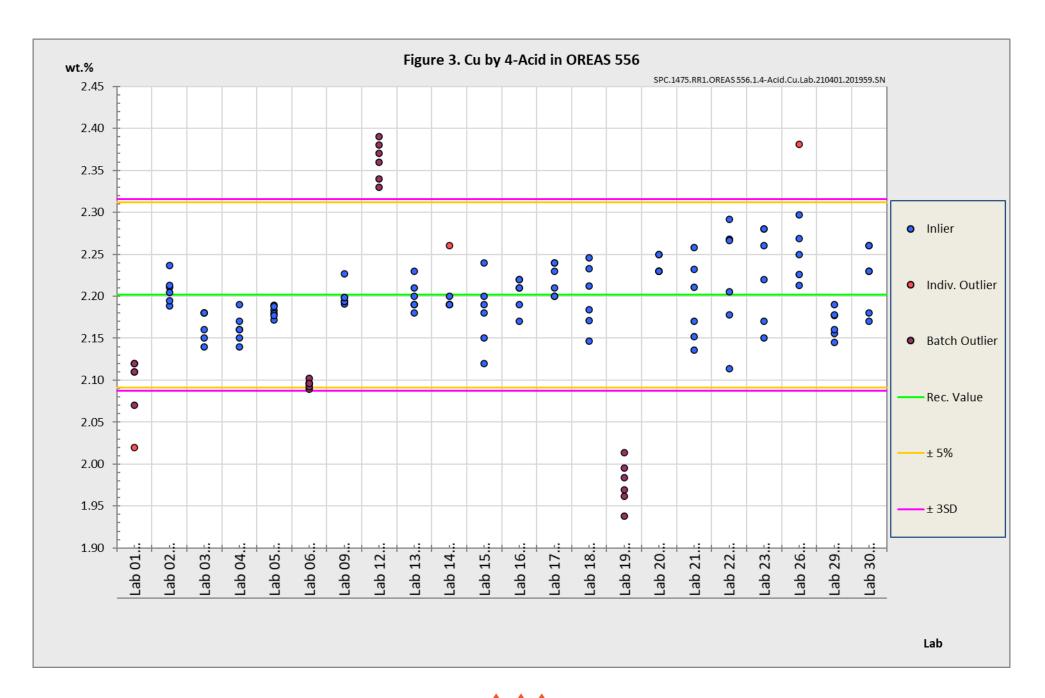
PARTICIPATING LABORATORIES

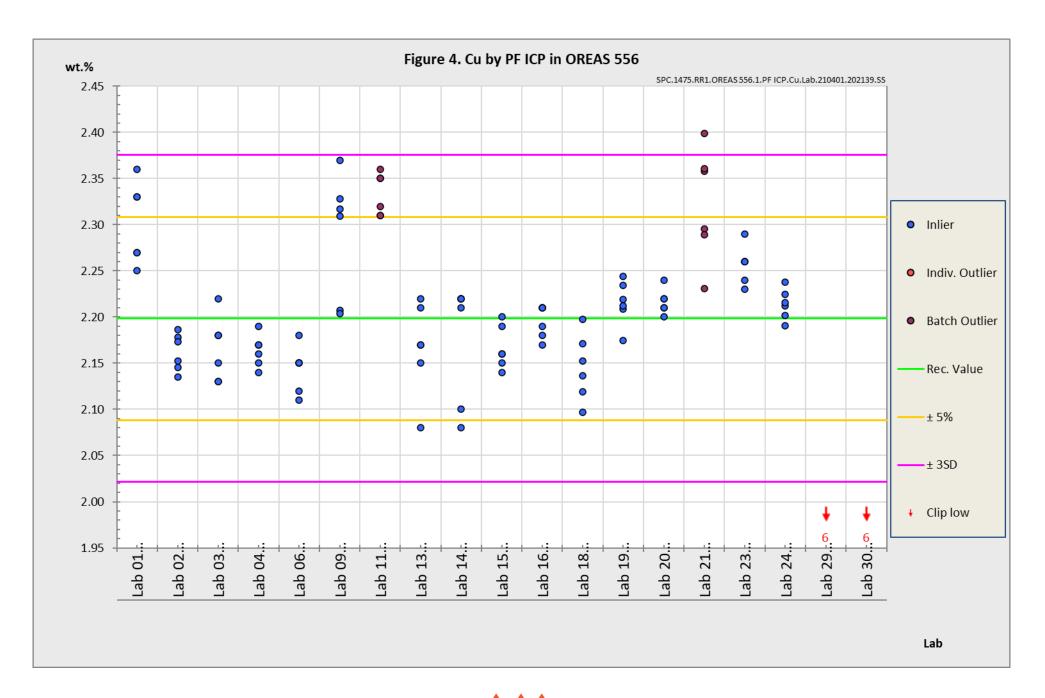
- 1. Actlabs, Ancaster, Ontario, Canada
- 2. AGAT Laboratories, Mississauga, Ontario, Canada
- 3. AH Knight, Kitwe, Copperbelt, Zambia
- 4. AH Knight, Lumwana, Mwinilunga, Zambia
- 5. ALS, Brisbane, QLD, Australia
- 6. ALS, Johannesburg, South Africa
- 7. ALS, Lima, Peru
- 8. ALS, Loughrea, Galway, Ireland
- 9. ALS, Vancouver, BC, Canada
- 10. American Assay Laboratories, Sparks, Nevada, USA
- 11. Bureau Veritas Commodities Canada Ltd, Vancouver, BC, Canada
- 12. Bureau Veritas Geoanalytical, Adelaide, SA, Australia
- 13. Bureau Veritas Geoanalytical, Perth, WA, Australia
- 14. Carsurin, Kendari, Sulawesi, Indonesia
- 15. Inspectorate (BV), Lima, Peru
- 16. Intertek Genalysis, Perth, WA, Australia
- 17. Intertek Testing Services Philippines, Cupang, Muntinlupa, Philippines
- 18. MSALABS, Vancouver, BC, Canada
- 19. Nagrom, Perth, WA, Australia
- 20. Ontario Geological Survey, Sudbury, Ontario, Canada
- 21. PT Geoservices Ltd, Cikarang, Jakarta Raya, Indonesia
- 22. PT Intertek Utama Services, Jakarta Timur, DKI Jakarta, Indonesia
- 23. Robinson International Afrique SAS, Lubumbashi, Katanga, Democratic Republic of the Congo
- 24. Ruashi Mine Site Laboratory, Lubumbashi, Katanga, Democratic Republic of the Congo
- 25. Saskatchewan Research Council, Saskatoon, Saskatchewan, Canada
- 26. SGS, Randfontein, Gauteng, South Africa
- 27. SGS Canada Inc., Vancouver, BC, Canada
- 28. SGS del Peru, Lima, Peru
- 29. SGS Geosol Laboratorios Ltda, Vespasiano, Minas Gerais, Brazil
- 30. UIS Analytical Services, Centurion, South Africa

Please note: To preserve anonymity, the above numbered alphabetical list of participating laboratories does not correspond with the Lab ID numbering on the scatter plots below.









PREPARER AND SUPPLIER

Certified reference material OREAS 556 was prepared and certified by:



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METROLOGICAL TRACEABILITY

The analytical samples were selected in a manner to represent the entire batch of prepared CRM. This 'representivity' was maintained in each submitted laboratory sample batch and ensures the user that the data is traceable from sample selection through to the analytical results that underlie the consensus values. Each analytical data set has been validated by its assayer through the inclusion of internal reference materials and QC checks during analysis.

The laboratories were chosen on the basis of their competence (from past performance in inter-laboratory programs undertaken by ORE Pty Ltd) for a particular analytical method, analyte or analyte suite, and sample matrix. Most of these laboratories have and maintain ISO 17025 accreditation. The certified values presented in this report are calculated from the means of accepted data following robust statistical treatment as detailed in this report.

Guide ISO/TR 16476:2016, section 5.3.1 describes metrological traceability in reference materials as it pertains to the transformation of the measurand. In this section it states, "Although the determination of the property value itself can be made traceable to appropriate units through, for example, calibration of the measurement equipment used, steps like the transformation of the sample from one physical (chemical) state to another cannot. Such transformations may only be compared with a reference (when available), or among themselves. For some transformations, reference methods have been defined and may be used in certification projects to evaluate the uncertainty associated with such a transformation. In other cases, only a comparison among different laboratories using the same method is possible. In this case, certification takes place on the basis of agreement among independent measurement results (see ISO Guide 35:2006, Clause 10)."

COMMUTABILITY

The measurements of the results that underlie the certified values contained in this report were undertaken by methods involving pre-treatment (digestion/fusion) of the sample. This served to reduce the sample to a simple and well understood form permitting calibration using simple solutions of the CRM. Due to these methods being well understood and highly effective, commutability is not an issue for this CRM. Being matrix-matched, OREAS 556 will display similar behaviour in the relevant measurement process to the routine 'field' samples for which OREAS 556 is designated to monitor. To maintain commutability, care should be taken to always ensure 'matrix matching' as close as practically achievable. The matrix and mineralisation style of the CRM is described in the 'Source Material' section and users should select appropriate CRMs matching these attributes to their field samples.

INTENDED USE

OREAS 556 is intended to cover all activities needed to produce a measurement result. This includes extraction, possible separation steps and the actual measurement process (the signal producing step). OREAS 556 may be used to calibrate the entire procedure by producing a pure substance CRM transformed into a calibration solution. OREAS 556 is intended for the following uses:

- For the monitoring of laboratory performance in the analysis of analytes reported in Table 1 in geological samples;
- For the verification of analytical methods for analytes reported in Table 1;
- For the calibration of instruments used in the determination of the concentration of analytes reported in Table 1.

STABILITY AND STORAGE INSTRUCTIONS

OREAS 556 has been prepared from a blend of copper-cobalt primary ore samples. It contains minor sulphide (1.69 wt.% S) and in its unopened state and under normal conditions of storage has a shelf life beyond ten years. Its stability will be monitored at regular intervals and purchasers notified if any changes are observed.

Single-use sachets

Following analysis of the CRM subsample it is the manufacturers' expectation that any remaining material is discarded. The stability of the material after opening the sachet is not within the scope of proper use. However, if opened sachets are resealed after opening, then under ordinary* storage conditions the CRM will have a shelf-life beyond ten years.

*ordinary storage conditions: means storage not in direct sunlight in a dry, clean, well ventilated area at temperatures between -5° and 50°C.

Repeat-use packaging (e.g. 500g plastic jars)

The stability of the CRM after opening the lid of the plastic jar is only affected by local atmospheric conditions with regard to oxidation and hygroscopic change. There is no segregation affect (please see our <u>Technical Note on Particle Segregation</u>).

The primary cause of change through oxidation is in relation to the breakdown of sulphide minerals to sulphates and is negligible for OREAS 556 given its sulphur concentration (1.69 wt.% S).

Hygroscopic change is the amount of absorbed moisture (weakly held H₂O- molecules on the surface of exposed material) following exposure to the local atmosphere. Usually, equilibration of material to the local atmosphere will only occur if the material is spread into a thin (~2mm thick) layer and left exposed for a period of 2 hours. OREAS 556 contains a non-hygroscopic matrix and therefore, exposure to a local atmosphere that is significantly different (in terms of temperature and humidity) from the climate during manufacturing will have negligible impact on the precision of results. The 'Physical Properties' section indicates the approximate moisture concentration.

COA-1475-OREAS 556-R1 Page: 23 of 26

INSTRUCTIONS FOR CORRECT USE

The certified values for OREAS 556 refer to the concentration level in its packaged state. It should not be dried prior to weighing and analysis.

500g jars permit repeated sampling as long as the lid is promptly re-secured to prevent airborne contamination.

Minimum sample size

As a practical guide, the minimum mass of sample used should match the typical mass that the laboratories used in the interlaboratory (round robin) certification program. This means that different sample masses should be used depending on the operationally defined methodology.

- Lithium borate fusion with X-ray fluorescence finish: ≥0.2g;
- Loss on Ignition (LOI) at 1000°C: ≥1g;
- C and S by infrared combustion furnace/CS analyser: ≥0.1g;
- Cu and Co by 5% sulphuric acid leach with ICP or AAS finish: 0.5g;
- Cu and Co by 10% sulphuric acid leach with ICP or AAS finish: ≥0.5g;
- Sodium peroxide fusion with ICP-OES and/or MS finish: ≥0.2g;
- 4-acid digestion with ICP-OES and/or MS finish: ≥0.25g;
- Agua regia digestion with ICP-OES and/or MS finish: ≥0.5g.

QC monitoring using multiples of the Standard Deviation (SD)

In the application of SD's in monitoring performance it is important to note that not all laboratories function at the same level of proficiency and that different methods in use at a particular laboratory have differing levels of precision. Each laboratory has its own inherent SD (for a specific concentration level and analyte-method pair) based on the analytical process and this SD is not directly related to the round robin program.

The majority of data generated in the round robin program was produced by a selection of world class laboratories. The SD's thus generated are more constrained than those that would be produced across a randomly selected group of laboratories. To produce more generally achievable SD's the 'pooled' SD's provided in this report include inter-laboratory bias. This 'one size fits all' approach may require revision at the discretion of the QC manager concerned following careful scrutiny of QC control charts.

The performance gates shown in Table 1 are intended only to be used as a first principle guide as to what a laboratory may be able to achieve. Over a period of time monitoring your own laboratory's data for this CRM, SD's should be calculated directly from your own laboratory's process. This will enable you to establish more specific performance gates that are fit for purpose for your application as well as the ability to monitor bias. If your long-term trend analysis shows an average value that is within the 95% confidence interval then generally there is no cause for concern in regard to bias.

It is important to note that in the analytical industry there is no standardisation of the aqua regia digestion process nor the 5% and 10% sulphuric acid leach process. These methods are partial empirical digests and differences in recoveries for various analytes are commonplace. These are caused by variations in the digest conditions and in the case for aqua regia digestion can include the ratio of nitric to hydrochloric acids, acid strength, temperatures, leach times and secondary digestions. Recoveries for sulphide-hosted base

metal sulphides approach total values, however, other analytes, in particular the lithophile elements, show greater sensitivity to method parameters. This can result in lack of consensus in an inter-laboratory certification program for these elements.

The approach applied here is to report certified values in those instances where reasonable agreement exists amongst a majority of participating laboratories. The results of specific laboratories may differ significantly from the certified values, but will, nonetheless, be valid and reproducible in the context of the specifics of the aqua regia method in use. Users of this reference material should, therefore, be mindful of this limitation when applying the certified values in a quality control program.

HANDLING INSTRUCTIONS

Fine powders pose a risk to eyes and lungs and therefore standard precautions such as the use of safety glasses and dust masks are advised.

LEGAL NOTICE

Ore Research & Exploration Pty Ltd has prepared and statistically evaluated the property values of this reference material to the best of its ability. The Purchaser by receipt hereof releases and indemnifies Ore Research & Exploration Pty Ltd from and against all liability and costs arising from the use of this material and information.

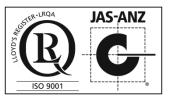
DOCUMENT HISTORY

Revision No.	Date	Changes applied
1	21 st April, 2021	Applied minor statistical correction to Cobalt by XRF method results.
0	7 th April, 2021	First publication.

QMS ACCREDITATION

ORE Pty Ltd is accredited to ISO 9001:2015 by Lloyd's Register Quality Assurance Ltd for its quality management system including development, manufacturing, certification and supply of CRMs.





CERTIFYING OFFICER

21st April, 2021

Craig Hamlyn (B.Sc. Hons - Geology), Technical Manager - ORE P/L



REFERENCES

Govett, G.J.S. (1983), ed. Handbook of Exploration Geochemistry, Volume 2: Statistics and Data Analysis in Geochemical Prospecting (Variations of accuracy and precision), P.O. Box 330, 1000 AH Amsterdam, The Netherlands.

ISO Guide 30:2015, Terms and definitions used in connection with reference materials.

ISO Guide 31:2015, Reference materials – Contents of certificates and labels.

ISO Guide 35:2017, Certification of reference materials - General and statistical principals.

ISO 16269:2014, Statistical interpretation of data – Part 6: Determination of statistical tolerance intervals.

ISO/TR 16476:2016, Reference Materials – Establishing and expressing metrological traceability of quantity values assigned to reference materials.

ISO 17025:2005, General requirements for the competence of testing and calibration laboratories.

Munsell Rock Color Book (2014). Rock-Color Chart Committee, Geological Society of America (GSA), Minnesota (USA).

APPENDIX

For Cu and Co by the two sulphuric acid leaches, specific methodologies were detailed for the participating laboratories to follow:

5% sulphuric acid leach

- 1. Weigh 0.500 ± 0.002g of sample pulp into a clean 250 ml flask.
- 2. Add to the flask 0.5g of Sodium Sulphite (AR Grade).
- 3. Add 50 ml of approximately 55 g/l Sulphuric acid solution (prepared from a 98% pure concentrated sulphuric acid).
- 4. Put the cap on the flask and start automatic shaking. Leave the sample on continuous shaking to leach for four (4) hours.
- 5. Remove the cap and add 25 ml of hydrochloric acid.
- 6. Dilute to a final volume of 250 ml with distilled/deionised water up to the mark and mix again by inverting at least 10 times. Allow the solution to settle for 30 minutes.
- 7. The solution is now ready to be analysed by ICP or AAS.

10% sulphuric acid leach

- 1. Weigh 0.500 ± 0.002g of sample pulp into a clean 250 ml flask.
- 2. Add to the flask 0.5g of Sodium Sulphite (AR Grade).
- 3. Add 50 ml of approximately 110 g/l Sulphuric acid solution (prepared from a 98% pure concentrated sulphuric acid).
- 4. Put the cap on the flask and start automatic shaking. Leave the sample on continuous shaking to leach for four (4) hours.
- 5. Remove the cap and add 25 ml of hydrochloric acid.
- 6. Dilute to a final volume of 250 ml with distilled/deionised water up to the mark and mix again by inverting at least 10 times. Allow the solution to settle for 30 minutes.
- 7. The solution is now ready to be analysed by ICP or AAS.